PROUTIAN PERIODIC TABLE: H - ATOM = BASIC BUILDING BLOCK

I Period –>	+16 H	II	+16 H	III	+48 H	IV	+48 H	V	+48 H	VI	+48 H	VII	GROUP
		Be 9		Mg 25		*Ge **73***		Sn 121		**Tm** 169		217	1a
		B 10		Mg 26		Ge 74		Sn 122		Er 170		218	1b
		B ***11***		**Al** 27		**As** 75		Sb 123		Yb 171		219	1c
		C 12		Si 28		Se 76		Sn 124		Yb 172		220	1d
		*C **13***		*Si **29***		Se 77		Te 125		*Yb **173***		***221***	1e
		N 14		Si 30		Se 78		Te 126		Yb/Lu174		Rn 222	1f
		N 15		***P 31***		*Br **79***		***I 127***		Lu 175		*Fr **223***	1g
		O 16		S 32		Se 80		Te 128		Yb 176		Ra 224	1h
H 1		*O **17***		S 33		Br 81		Xe 129		Hf 177		225	2a
*H **2***		O 18		S 34		Kr/Se 82		Te 130		Hf 178		Ra 226	2b
*He **3***		***F 19***		Cl 35		*Kr **83***		Xe***131***		Hf ***179***		*Ac **227***	2c
He 4		Ne 20		Ar 36		Kr 84		Xe 132		Hf 180		Ra 228	2d
		Ne 21		*Cl **37***		Rb 85		**Cs** 133		Ta ***181***		***229***	2e
Li 6		Ne 22		Ar 38		Kr/Sr 86		Xe 134		W 182		230	2f
*Li **7***		***Na 23***		*K* 39		Rb/Sr 87		Ba 135		Re/W 183		Pa 231	2g
		Mg 24		Ca/Ar 40		Sr 88		Ba 136		W 184		**Th** 232	2h
				*K **41***		***Y 89***		*Ba **137***		Re 185		***233***	3a
				Ca 42		Zr 90		Ba 138		W 186		Pa 234	3b
				*Ca **43***		Zr 91		***La 139***		*Re* 187		235	3c
				Ca 44		Zr/Mo 92		Ce 140		Os 188		236	3d
				Sc 45		**Nb** 93		**Pr** 141		Os 189		Np 237	3e
				Ti 46		NbMoZr94		Nd 142		Os 190		U/Pu238	3f
				*Ti **47***		Mo 95		Nd 143		*Ir **191***		***239***	3g
				Ti 48		Mo 96		Nd/Sm144		Os 192		Pu 240	3h
				Ti 49		*Mo **97***		Nd 145		*Ir **193***		***241***	4a
				Ti/Cr 50		Mo 98		Nd/Sm146		Pt 194		242	4b
				V 51		Ru 99		Sm 147		Pt 195		Am 243	4c
				Cr 52		Mo/Ru 100		Sm/Nd148		Pt 196		Cm 244	4d
				*Cr **53***		*Ru **101***		*Sm **149***		***Au 197***		245	4e
				Fe/Cr 54		Ru 102		Sm 150		Pt/Hg198		Cm 246	4f
				Mn 55		***Rh 103***		*Eu **151***		*Hg **199***		*Bk 247*	4g
				Fe 56		Pd/Ru104		Sm 152		Hg 200		248	4h
				Fe 57		Pd 105		Eu 153		Hg 201		239	5a
				Ni 58		Pd 106		Sm 154		Hg 202		Cf 250	5b
				Co 59		*Ag **107***		Gd 155		Tl 203		***251***	5c
				Ni 60		Pd 108		Gd 156		Pb 204		Cf 252	5d
				*Ni **61***		*Ag **109***		*Gd **157***		Tl 205		253	5e
				Ni 62		Pd/Cd 110		Gd 158		Pb 206		254	5f
				Cu 63		Cd 111		**Tb** 159		Pb 207		255	5g
				Zn 64		Cd 112		Gd/Dy 160		Pb/Po208		256	5h
				Cu 65		*Cd/In **113***		Dy 161		***Bi 209***		*Fm **257***	6a
				⇓		⇓		⇓		⇓		⇓	⇓
				Ge 72		Sn 120		Er 168		Bh 216		264	6h

Legend

C 12 ...Carbon: 12 hydrogen (H-) constituents.
Period: +16H or +48H = building blocks

*B **11**...*.prime number.
Be 9...relative abundance: 100.

Ar 40 and Ca 40 are ***isomers:*** 40 H-constituents each but different shape! Therefore Ar 40 and Ca 40 have different 'inert' masses (drag coefficients).

Not all isomers are mentioned.

Construction schedule for noble gases:
Ne20 + 64H = Kr84;
Ne20 + 20H = Ar40.
(Ar36: only 0.34 relative abundance)